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the exercises within it, and answering the questions at the end, you should be able to: Describe the nature of atoms and their constituents, particularly the behaviour of electrons and the formation of ions. Apply your understanding of

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bonding within
minerals.

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> Across. The color of a mineral in a powdered form. The prefix tetra means this. An element that has different forms of that element based on different

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Minerals of neutrons. The characteristic of minerals that are orderly and repetitive. These subatomic particles are located in every energy level ...

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Volcano World

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I.Main Body. 1.

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Building Blocks of
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STUDY. PLAY.

atoms. tiny
particles that make
up matter;
composed of
protons, electrons,
and neutrons.

mineral. a naturally
occurring, inorganic
solid with a definite
chemical
composition and an
orderly

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arrangement of atoms. crystal. a solid in which the atoms are arranged in orderly, repeating patterns.

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characteristics

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2.6 Mineral

Properties Minerals

are universal. A

crystal of hematite

on Mars will have

the same properties

as one on Earth,

and the same as

one on a planet

orbiting another

star. That ' s good

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Minerals news for geology students who are planning interplanetary travel since we can use those properties to help us identify minerals ...

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properties of
minerals based on
chemical

composition and
structure, and
importance of

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"How Does Earth
Work) by Gary
Smith and Aurora
Pun.

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Minerals; Linda C.

- 31 cards. What is a rock? A naturally occurring

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Minerals combination of one or more materials. Minerals. Composed of crystals which are orderly arrangements of atoms bonded together in geometric patterns. How Minerals form ...

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FERTILIZERS AND
AGRICULTURAL.
Dig a little deeper
All living things
need minerals.

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and Minerals

Volcano World.

Observe common
objects made of
minerals ClassZone.

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about how the arrangement of atoms in minerals affects their physical properties.
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identifying different
igneous rocks.

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2.3: Mineral
Groups: Minerals
are grouped
according to the
anion part of their
formula, with some

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Common types
being oxides,
sulphides,
sulphates, halides,
carbonates,
phosphates,
silicates, and native
minerals. 2.4:

Silicate Minerals:
Silicate minerals
are, by far, the
most important
minerals in Earth ' s
crust.

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| Physical Geology

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2.3 Mineral Groups

Most minerals are made up of a cation (a positively charged ion) or several cations, plus an anion (a negatively charged ion (e.g., S^{2-})) or an anion complex

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(e.g., SO_4^{2-}). For example, in the mineral hematite (Fe_2O_3), the cation is Fe^{3+} (iron)

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